Effect of cholestyramine on the absorption of vitamin D₃ and calcium

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SUMMARY The absorption of 10 μg doses of vitamin D₃-3H in rats was found to be decreased by the addition of cholestyramine to the diet in an amount sufficient to cause steatorrhoea. Studies in vitro suggested that the effect of cholestyramine was due to its known ability to bind bile salts and thus disrupt micelles containing vitamin D.

The absorption of 1 mg calcium chloride was similar in both control and cholestyramine-fed rats, whether estimated by the whole body counting or the faecal excretion technique.

These findings show that the administration of cholestyramine causes a reduction in the absorption of vitamin D₃-3H but not of radiocalcium.

Cholestyramine is an anion exchange resin which binds bile acids in the intestinal lumen, thus impairing their absorption. This property has led to its use in the treatment of hypercholesterolaemia (Hashim and Van Itallie, 1965; Howard, Brusco, and Furman, 1966) and the pruritus of partial biliary obstruction (Carey and Williams, 1961; Datta and Sherlock, 1963). In addition, cholestyramine has been shown to induce steatorrhoea in both rats (Harkins, Hagerman, and Sarett, 1965) and man (Hashim, Bergen, and Van Itallie, 1961).

Vitamin D deficiency after therapy with cholestyramine has not been reported, although malabsorption of this fat-soluble vitamin might be expected to accompany cholestyramine-induced steatorrhoea. This, if sufficiently marked to cause vitamin D deficiency, might eventually lead to an impairment of calcium absorption. On the contrary, Briscoe and Ragen (1963) have observed increased calcium absorption in three patients treated with cholestyramine, and there have also been reports of calcification in the biliary tree of patients receiving this drug on a long-term basis (Schaffner, Klion, and Latuff, 1965; Wells, Knapshield, and Davis, 1968).

In an attempt to clarify this apparent paradox the effect of cholestyramine on the absorption of both radioactive vitamin D and calcium has been measured in rats. The results of these studies show that cholestyramine markedly interferes with the absorption of vitamin D₃-3H but not with that of radiocalcium. In addition, some studies in vitro were undertaken to determine the mechanism of the interaction between cholestyramine, bile acids, and vitamin D.

METHODS

PREPARATION AND PURIFICATION OF VITAMIN D₃-3H
Crystalline vitamin D₃ was labelled with tritium by random exchange and then purified and characterized as previously described (Thompson, Lewis, and Booth, 1966). The batch of vitamin D₃-3H used in these experiments had a specific activity of 62.8 μc per mg, and a biological activity of approximately 40 iu per μg when assayed in rachitic rats (Bourdillon, Bruce, Fischmann, and Webster, 1931). The stock solution of vitamin D₃-3H was stored at 4°C and repurified at approximately weekly intervals by thin-layer chromatography. The labelled vitamin was chromatographed on plates of Kieselgel H₄ containing Rhodamine 6G with chloroform as the solvent, and its position located with an ultraviolet lamp. The vitamin was then eluted with chloroform and its specific activity determined.

ABSORPTION OF VITAMIN D₃ Eleven male Wistar rats, each weighing approximately 200 g, were divided into two groups. All rats received a standard diet containing 3% fat and 5% fibre, but in one group cholestyramine (Merck, Sharp, and Dohme) was mixed in with the
diet in a proportion of 4% of dry weight, equivalent to 2% of wet weight. After nine days of equilibration on this regime the rats were put in metabolic cages and their vitamin D absorption was determined as follows.

Doses of 10μg (400iu) of vitamin D3-2H in chloroform were dried in a rotary evaporator and then redissolved in 0-2 ml propylene glycol in ethanol 1:1, v/v. Calibrated doses were administered via an intragastric tube to non-fasting rats. Faeces were collected in individual containers for five days after the vitamin D3 administration and then weighed and homogenized in 100 ml water. Approximately 5 g was then weighed and refluxed for two hours in 100 ml 2:1 chloroform-methanol, v/v. The faecal extract was cooled and filtered into separating funnels. The flask was rinsed with 10 ml 2:1 chloroform-methanol, v/v, and the phases were separated with 22 ml water (Folch, Lees, and Sloan Stanley, 1957). Centrifugation was employed to speed separation. The chloroform extract was dried down in a rotary evaporator and then redissolved in 5 ml toluene, of which 0-5 ml was added to 10 ml of a scintillation fluid containing butyl PBD (2-(4-butylphenyl)-5-(4-biphenyl)-1, 3,4-oxidiazole) 6 g/l.1 Radioactivity in these samples was assayed in a Beckman model 250 B liquid scintillation counter. Quenching was corrected by means of an external standard.

Samples of 0-5 ml of both urine and of the methanol phase of the faecal extract were counted in 10 ml of a scintillation fluid containing butyl PBD 8 g, naphthalene 80 g, ethoxyethanol 400 ml, and toluene 600 ml. Faecal fat determinations were carried out by the method of van de Kamer, ten Bokkel Huinink, and Weyers (1949).

STUDIES IN VITRO Approximately 40 μg of vitamin D3-2H in chloroform was dried down, either alone or together with appropriate quantities of glyceryl-1-monooleate6 and linoleic acid.8 In some experiments linoleic acid-14C4 was used to label the fatty acid. The dried lipid was gradually dissolved in 8ml of 40 mM sodium taurocholate8 with heating and manual agitation of the flask. In some experiments sodium taurocholate-2414C8 was also added. Finally 8 ml of a saline-phosphate buffer (0-15 M sodium chloride: 0-1 M sodium dihydrogen phosphate, 5:1 v/v) was added to dilute the final concentration of taurocholate to 20 mM, and the pH was adjusted to 6-7 with 1 N sodium hydroxide. By this means micellar solutions were prepared to contain 6-5 μM of vitamin D3-2H in either 20 mM taurocholate alone, or with added 4-8 μM glyceryl monooleate plus 9-6 μM of linoleic acid per milliliter.

Weighed amounts of cholestyramine were placed in stoppered test tubes and wetted with 0-5 ml of the saline-phosphate buffer. Of each test solution, 5 ml was carefully pipetted into triplicate into the tubes containing 25, 50, and 100 mg of cholestyramine and 0-1 ml was taken immediately for radioassay. After 15 minutes of vigorous shaking at 37°C in a water bath, the tubes were centrifuged at 3,000 × g for five minutes and a further 0-1 ml of the supernatant was then taken for counting. There was no change in pH during the entire procedure. The difference in radioactivity (total DPM) in the solution before mixing and after centrifugation was taken as the amount of vitamin D3-2H or linoleic acid-14C4 or sodium taurocholate-2414C4 removed from solution by the action of cholestyramine. In the double isotope experiments 2H and 14C were counted in separate channels and appropriate corrections made for the spillover of 14C into the 2H channel.

Of the mixed micellar solution, 1 ml taken before the addition of cholestyramine and 1 ml of the supernatant taken after centrifugation were each added to 20 ml 2:1 v/v chloroform-methanol in separating funnels. The phases were separated with 4 ml of water. The chloroform phase was dried down, and then chromatographed together with a vitamin D3 marker, as described above. For radioassay, 1 cm strips of the chromatograph were scraped off the plate into counting vials containing 10 ml scintillation fluid.

RADIOCALCUM ABSORPTION Radioactive calcium (47Ca), freshly prepared by the Radio-chemical Centre, was mixed with 1 mg of nonradioactive calcium chloride. Absorption was measured both by means of the faecal excretion of radioactivity and by whole body counting. Two similar studies were carried out. The first consisted of the 11 rats previously given vitamin D3-2H, six of whom had continued to receive 4% cholestyramine during the previous three weeks. The second group consisted of 12 fresh rats who received 2.5 μg (100 IU) of unlabelled vitamin D3 by gastric tube before the experiment was started, so as to avoid any possible bias that might result from malabsorption of vitamin D in cholestyramine-treated animals. Six of the second group of rats subsequently received 4% cholestyramine in their diet for three days. All 12 were then placed in individual metabolic cages and starved overnight until two hours before the administration of calcium, at which time they resumed their diet. The results within each of the two control and two cholestyramine subgroups were found to be very similar and were therefore pooled.

Alternate control and treated animals were given 0.3 ml of the labelled calcium chloride containing 1 to 3 μC by stomach tube at 15-minute intervals. Half an hour after dosage, each rat was placed in a polythene bag in a beaker and lowered into a large ring counter. Radiation from 47Scandium, the main daughter product of 47Ca, was excluded by a lead shield. Counting proceeded until 10,000 counts had accumulated above the background. Exactly 24 hours later the animal was killed and the entire intestinal tract removed. The carcass was then recounted, with appropriate correction for the 24-hour decay in 47Ca activity.

Control doses, urine, faeces, and the small and large bowel of each rat were placed in separate counting vials and counted in an automatic gamma counter (Nuclear Enterprises, Edinburgh, Scotland). Samples from control and treated animals were alternated to minimize the error due to isotopic decay.

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1Ciba Ltd. (ARL), Duxford, Camb.
2Calbiochem, Calif.
3Koch-Light, London.
4Radio-chemical Centre, Amersham.
5Maybridge, Tintagel, Cornwall.
6Tracerlab, Waltham, Mass.
RESULTS

ABSORPTION OF VITAMIN D₃₃H The mean body weight, food intake, and faecal weight of the control and cholestyramine-treated rats were all very similar (Table I). However, the faecal fat excretion of the cholestyramine group (2·86 ± 0·29 m-equiv/day) was more than double that of the controls (1·23 ± 0·05 m-equiv/day; p<0·001).

<table>
<thead>
<tr>
<th>Group</th>
<th>No. of Rats</th>
<th>Body Weight (g)</th>
<th>Food Intake (g/day)</th>
<th>Faecal Weight (g/day)</th>
<th>Faecal Fat (m-equiv/day)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td>5</td>
<td>190 ± 23¹</td>
<td>47 ± 2</td>
<td>33 ± 7</td>
<td>1·23 ± 0·05</td>
</tr>
<tr>
<td>Cholestyramine-treated</td>
<td>6</td>
<td>201 ± 19</td>
<td>47 ± 2</td>
<td>37 ± 5</td>
<td>2·86 ± 0·29</td>
</tr>
</tbody>
</table>

¹Results are mean ± 1 SD

The effect of cholestyramine on the absorption of vitamin D₃₃H is shown in Table II. The faecal excretion of radioactivity was measured during the five days following the oral dose of vitamin D₃₃H. Absorption was calculated as the dose of radioactivity given minus the chloroform-soluble radioactivity excreted in the faeces, expressed as a percentage of the administered dose. The absorption of vitamin D₃₃H in the control group was 64·7 ± 4·1% whereas it was reduced to 28·1 ± 14·7% in the cholestyramine-treated animals (p<0·001). In addition there was also a difference between the two groups regarding the methanol-soluble faecal radioactivity, which amounted to 7·4 ± 2·7% in the controls but was only 1·9 ± 1·0% in the cholestyramine group (p<0·01; see Table II).

<table>
<thead>
<tr>
<th>Group</th>
<th>No. of Rats</th>
<th>Faecal Radioactivity (% of Dose¹)</th>
<th>Absorption (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Chloroform Fraction</td>
<td>Methanol Fraction</td>
</tr>
<tr>
<td>Control</td>
<td>5</td>
<td>35·3 ± 4·1¹</td>
<td>74·2 ± 27</td>
</tr>
<tr>
<td>Cholestyramine-treated</td>
<td>6</td>
<td>71·0 ± 14·7</td>
<td>28·1 ± 14·7</td>
</tr>
</tbody>
</table>

¹Results are mean ± 1 SD

Urinary radioactivity never exceeded 1·2% of the administered dose during the subsequent three days in either group.

FIG. 1. Amount of vitamin D₃₃H removed from taurocholate and mixed micellar solutions by 25, 50, and 100 mg cholestyramine. This is calculated as the percentage difference between the radioactivity initially present minus the residual radioactivity after the addition of cholestyramine.

EFFECT OF CHOLESTYRAMINE ON VITAMIN D₃ Figure 1 illustrates the comparative effects of 25, 50, and 100 mg of cholestyramine in removing vitamin D₃₃H from solution in mixed micelles and taurocholate micelles.

Centrifugation of solutions of vitamin D₃₃H in either taurocholate or mixed micelles caused no change in radioactivity. However, the addition of 25, 50, and 100 mg of cholestyramine to the mixed micellar solution resulted in the disappearance from the supernatant solution respectively of 41·2%, 70·4%, and 72·2% of the radioactivity initially present. In contrast 100% of the vitamin D₃₃H in the 20 mm taurocholate solution was removed by 100 mg of cholestyramine. This amount of cholestyramine was equivalent to 2% w/v, which was the same concentration as was used in the studies in vivo.

In another experiment equal amounts of vitamin D₃₃H were dissolved in the following media: 20 mm taurocholate, taurocholate plus 4·8 μm/ml glyceryl monooleate, taurocholate plus 9·6 μm/ml linoleic acid and mixed micelles. Cholestyramine 100 mg was then added to each of these solutions, as previously described, and the radioactivity remaining in each supernatant was determined. The results are shown in Figure 2. It is evident that less vitamin D was removed from the solutions containing monoolein (39·4%) or mixed micelles (48·9%).
than from those containing fatty acid (83·2%) or taurocholate alone (100%).

It was shown on thin-layer chromatography that 83% of the radioactivity present in the mixed micellar solution before the addition of cholestyramine had the same R$_t$ as vitamin D$_3$. After admixture with cholestyramine and centrifugation, 75% of the residual radioactivity was located in this region.

In a subsequent study linoleic acid-1-14C was used to label the fatty acid component of the mixed micellar solution. In this instance it was found that 62·4% of the vitamin D$_3$-3H and 71·1% of the linoleic acid-1-14C were removed from solution by 100 mg of cholestyramine.

**Effect of Cholestyramine on Taurocholate-14C**

In order to investigate the varying susceptibility to cholestyramine of vitamin D$_3$-3H when incubated in different forms of micellar solution the following experiment was performed.

Sodium taurocholate-24-C$_{14}$ was added to the taurocholate component of pure bile salt micelles, monoglyceride micelles, fatty acid micelles, and mixed micelles. The results (Table III) showed that the amount of taurocholate-14C removed from each micellar solution by 100 mg cholestyramine was similar, irrespective of the type of micelle, and ranged from 70·3% for taurocholate alone to 63·8% for linoleic acid micelles.

**Radio Ca absorption** The absorption of $^{47}$Ca in the two groups of rats is shown in Table IV.

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![Graph](image-url)

**Figure 2.** Amount of vitamin D$_3$-3H removed from various micellar solutions by cholestyramine 100 mg. TC = taurocholate 20 mM, LA = linoleic acid, MO = monoglyceride.

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**Table III**

<table>
<thead>
<tr>
<th>Type of Micelle</th>
<th>Taurocholate-14C Removed (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Taurocholate (TC)</td>
<td>70·3</td>
</tr>
<tr>
<td>Fatty acid (TC + FA)</td>
<td>63·8</td>
</tr>
<tr>
<td>Monoglyceride (TC + MO)</td>
<td>66·3</td>
</tr>
<tr>
<td>Mixed (TC + FA + MO)</td>
<td>67·0</td>
</tr>
</tbody>
</table>

Using the whole body counting method absorption was 50·2 ± 13·8% in the control group and 48·4 ± 6·2% in the cholestyramine-treated group.

Measurement of the faecal excretion of $^{47}$Ca showed the absorption to be 44·8 ± 13·1% in controls and 43·2 ± 14·4% in the cholestyramine groups. These results suggest that radio-calcium absorption in the two groups was very similar. However, when the control and cholestyramine results were pooled and the two methods of estimating calcium absorption compared, no significant correlation between them could be shown ($r = 0·32$, $p > 0·05$).

In both groups of rats less than 1% of administered radioactivity was recovered from urine within 24 hours.

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**Table IV**

<table>
<thead>
<tr>
<th>Group</th>
<th>No. of Rats</th>
<th>Weight (g)</th>
<th>Absorption of 1 mg of $^{47}$Calcium (% of dose)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td>Whole Body Method</td>
</tr>
<tr>
<td>Control</td>
<td>11</td>
<td>238 ± 66</td>
<td>50·2 ± 13·8</td>
</tr>
<tr>
<td>Cholestryramine treated</td>
<td>12</td>
<td>235 ± 62</td>
<td>48·4 ± 6·2</td>
</tr>
</tbody>
</table>

**Discussion**

These studies clearly show that the administration of cholestyramine to rats in a dose sufficient to produce steatorrhoea leads to marked malabsorption of vitamin D$_3$. Harkins et al (1965) found that the degree of steatorrhoea induced was proportional to the dose of cholestyramine given. In the present study cholestyramine was administered as 4% of the dry weight of diet, since this was considered to be the minimal amount required to induce consistent steatorrhoea.

A strictly physiological dose of vitamin D is difficult to determine, but is probably of the order of 0·25 μg or 10 iu daily (Carlsson and Lindquist, 1955). In the present study, owing to the limited
specific activity of the vitamin D$_3$-3H, the minimal
dose which could be used was 10 µg. However,
Schachter, Finkelstein, and Kowarksi (1964) found
no significant difference in the percentage absorption
of 40 µg compared with 1 mg of vitamin D in intact
rats, which suggests that vitamin D absorption is
not dose-dependent within this range.

The radioactivity recovered in the chloroform
phase of faecal extracts following administration of
vitamin D$_3$-3H probably represented both un-
absorbed vitamin D$_3$ and biologically inactive
metabolites formed by the action of colonic bacteria
In contrast, methanol-soluble radioactivity was
probably largely due to biliary metabolites, formed
by conjugation in the liver. In the cholestyramine-
treated rats an increased faecal excretion of chole-
form-soluble radioactivity was associated with
reduced amounts of methanol-soluble radioactivity,
suggesting that the amount of radioactivity excreted
in bile was dependent upon the amount of vitamin
absorbed. As has been shown previously (Thompson
et al, 1966), the urinary excretion of radioactivity
after administration of vitamin D$_3$-3H was minimal.

Bile salts are necessary for the absorption of
vitamin D (Schachter et al, 1964), and the mal-
absorption of vitamin D$_3$-3H in the cholestyramine-
treated rats presumably resulted from binding of
bile salts by the resin and their subsequent excretion
in the faeces (Carey and Williams, 1961; Tennent,
Overall depletion of bile salts due to interruption of
the enterohepatic circulation (Weiner and Lack,
1968) would tend to impair micelle formation, but
local binding of bile salts in vitamin D$_3$ containing
micelles is probably also an important factor. This
latter effect of cholestyramine was illustrated by the
studies in vitro which showed that over 60% of the
vitamin D in mixed micelles was removed from
solution by the resin. When the vitamin D was
dissolved in taurocholate micelles it appeared to be
even more susceptible to the action of chole-
styramine, which caused its total disappearance from
solution (Fig. 1). Although Gallo, Bailey, and
Sheffner (1965) found that unionized drugs such as
digoxin may be non-specifically bound to chole-
styramine, it seems unlikely that the resin would bind
vitamin D$_3$ to a sufficient extent to explain the present
findings.

The fatty acid component of mixed micelles was
removed from solution by cholestyramine to a
similar extent as vitamin D, although whether this
was due to direct binding of ionized fatty acid was
not determined. Vitamin D appeared to be least
susceptible to the effect of cholestyramine when
dissolved in a micellar solution of glyceryl mono-
oleate. This protective effect of monoglyceride for
vitamin D was also evident when using mixed
micelles. Since the amount of taurocholate re-
moved from solution by cholestyramine was
fairly constant, whatever the type of micelle, it
seems probable that the beneficial effect of mono-
glyceride was related to its ability to remain in
micellar solution even after the resin had mark-
edly reduced the concentration of taurocholate.
Hofmann (1963) has shown that monooleate is
solubilized on almost a mole for mole basis by
taurodeoxycholate, and in the present study the
residual concentration of taurocholate was in the
region of 5 to 7 mm. This concentration would
theoretically have been able to maintain in micellar
form virtually all of the monooleate originally
present and consequentially much of the vitamin D.

Malabsorption of fat-soluble vitamins due to
cholestyramine therapy has not been proven in man
although Visintine, Michaels, Fukayama, Conklin,
and Kinsell (1961) reported the development of
vitamin K deficiency in a patient on long-term
cholestyramine therapy. However, the patient
described had primary biliary cirrhosis so that
vitamin K malabsorption could not be ascribed
solely to the resin, although the work of Whiteside,
Harkins, Fluckiger, and Sarett (1965) suggested that
it would be a contributory factor. The present studies
suggest the possibility that vitamin D deficiency
could arise during long-term cholestyramine
therapy: regular parenteral administration of
vitamin D, and probably the other fat-soluble
vitamins, would seem advisable in patients receiving
the resin in amounts sufficient to induce steatorrhea.
This problem is particularly relevant in patients
with chronic biliary obstruction, in whom vitamin D
absorption is often impaired, even before the admin-
istration of cholestyramine (Thompson et al, 1966).

The effect of long-term administration of chole-
styramine on calcium absorption has previously
been investigated by Briscoe and Ragan (1963).
These authors found an increase in the absorption
of calcium, given with a meal, in the three human
subjects whom they studied. Their findings are
somewhat difficult to reconcile with the existing
experimental evidence, since Weibling and Hold-
sworth (1966) and Lengemann and Dobbins (1958),
both working with rats, showed that bile salts
enhanced the absorption of calcium. This effect could
be abolished by ligating the common bile duct and
it seems reasonable to assume that cholestyramine
might act in an analogous manner. However, this
was not borne out by the results of the present
study, which showed that administration of chole-
styramine over periods of three days to three weeks
had no effect on radiocalcium absorption. There
were wide individual variations in radiocalcium absorption in both control and cholestyramine-treated rats, as has been found by others (Hertoghe, Phinney, and Rubin, 1967), but there were no significant differences in mean absorption between the two groups, irrespective of whether absorption was calculated from total body uptake or from the faecal excretion of \(^{47}\text{Ca}\).

Hertoghe et al (1967) found good agreement between the mean absorption of \(^{47}\text{Ca}\) estimated by the whole body counting and faecal excretion methods, although they did not analyse their data statistically. They used young rats of 80 to 100 g, on account of their reproducibly high absorption and retention of radiocalcium. However, when the pooled results obtained by each method during the present study were compared, no significant correlation between them could be shown. The individual results obtained by the whole body method were usually higher than those obtained by the faecal excretion method, with less variation about each mean. These findings suggest that the whole body counting method was possibly the more reliable.

REFERENCES


